

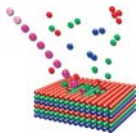
Ion beam surface engineering for highly active nanocatalysts

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Ion Beam Bombardment

application to Diesel Oxidation catalysts (DOC)



- fragment and disperse the metal in the form of nanoparticles with well-defined characteristics
- enhance the stability and the resistance to aging
- significant reduction of the operating temperature



Ion beam implanter

N⁺ ion beam at 52.5 keV
 current density of 3.75 $\mu\text{A}/\text{cm}^2$

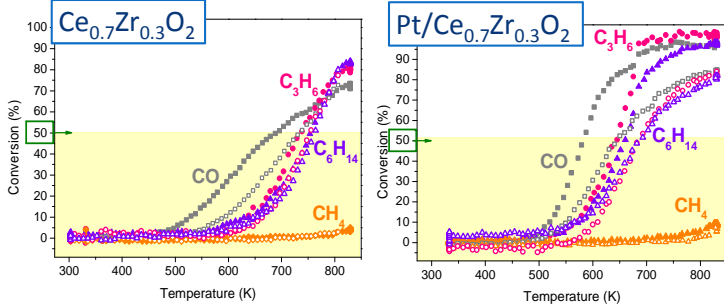
Materials

- $\text{Ce}_{0.7}\text{Zr}_{0.3}\text{O}_2$
- 1%Pt-Pd/ $\text{Ce}_{0.7}\text{Zr}_{0.3}\text{O}_2$



TPO Temperature Programmed Oxidation

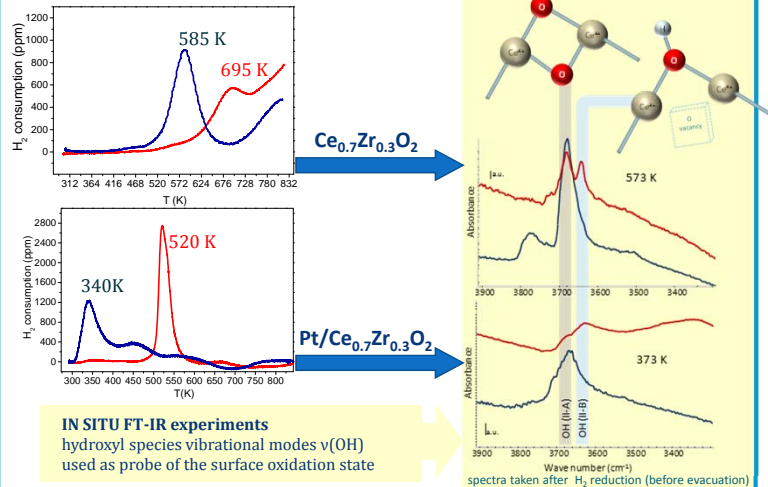
10 % O₂, 2000 ppm CO, CH₄, C₃H₆, C₆H₁₄ in Ar from r.t. to 823K, 2K/min



| | Ce _{0.7} Zr _{0.3} O ₂ | | Pt/Ce _{0.7} Zr _{0.3} O ₂ | |
|--|--|-----------|---|-----------|
| | non treated | bombarded | non treated | bombarded |
| T ₅₀ CO (K) | 739 | 696 | 640 | 579 |
| T ₅₀ C ₃ H ₆ (K) | 756 | 738 | 682 | 645 |
| T ₅₀ C ₆ H ₁₄ (K) | 764 | 754 | 693 | 660 |
| Conversion % CH ₄ @550°C | 4.2 | 4.9 | 8.4 | 9.4 |

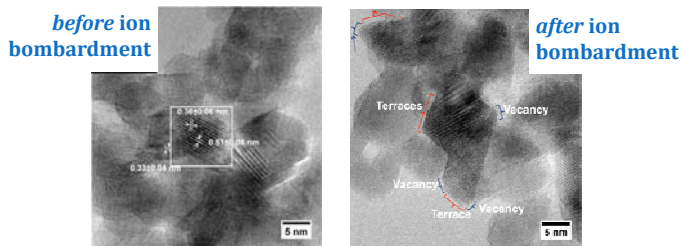
TPR Temperature Programmed Reduction

4000 ppm H₂ in Ar from r.t. to 823K, 5K/min



IN SITU FT-IR experiments
 hydroxyl species vibrational modes $\nu(\text{OH})$
 used as probe of the surface oxidation state

HRTEM Pt/Ce_{0.7}Zr_{0.3}O₂

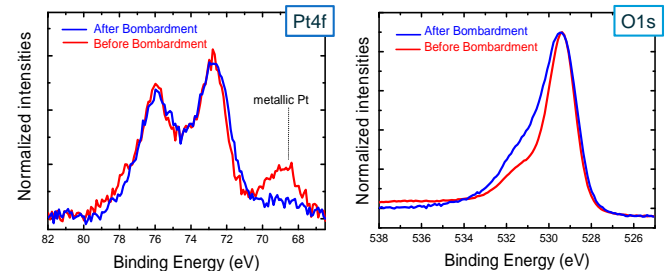


- highly faceted **nanoparticles** before and after the ion treatment
- upon ion bombardment: increment on the surface of defective units. Atomic **vacancies** are visible at the corner of the nanoparticles and extended **terraces**

the presence of **vacancies** and **terraces** is one of the driving causes for the enhanced catalytic activity of nanoscale systems

XPS Pt/Ce_{0.7}Zr_{0.3}O₂

EFFECT ON THE ELECTRONIC PROPERTIES



- heterogeneous **charge effect** on catalyst powder after bombardment under the X-Ray excitation
- the global composition of the catalyst is maintained and **novel structure** changes of the active sites.
- the oxygen peak is modified in its high binding energy part
 → **oxygen chemical environment is modified** by the bombardment

Conclusions: Effects of Ion Bombardment

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- Formation of **morphologic and structural defects on the surface** upon bombardment (vacancies & terraces)
- Creation of preferential absorption sites and routes for oxygen (increasing the **oxygen activation and mobility**)
- Modification of the local **electronic structure of the sites** (localised charges), affecting both metal and support

- Enhanced reducibility** of both CZ and Pt/CZ bombarded materials
- Enhanced catalytic activity** of nanoscale systems
- Better grafting of metal particles, i.e. better **resistance to ageing**

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