Nanocatalysts from Ionic Liquid Precursors for the Direct Conversion of CO₂ to Hydrocarbons

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Introduction

The direct conversion of carbon dioxide (CO₂) into hydrocarbons in the gasoline range (C₅-C₁₁) is a highly desirable process as a sustainable production route and it provides a key solution to managing the current CO₂ waste emissions. The reaction proceeds *via* two main consecutive reactions: Reverse Water Gas Shift (RWGS) to produce CO followed by the further conversion of CO to hydrocarbons *via* the Fischer–Tropsch reaction². This process is achieved by a multifunctional iron-based catalyst supported on zeolites providing three types of active sites (Fe₃O₄, Fe₅C₂ and acid sites), which cooperatively catalyse a tandem reaction¹.

To date, attempts at synthesising a suitable catalyst for the direct hydrogenation reaction follow a conventional precipitation procedure, whereby Iron Oxide Nanoparticles (*IONs*) are produced and then embedded within a zeolite structure by granule mixing. This method provides no control over the size and shape of the IONs formed; a characteristic of imperative importance due to its significant effect on the hydrocarbon product distribution obtained. In our novel approach, ionic liquids are utilised for the synthesis of the *IONs* resulting in better control over size and morphology of the nanostructured material, and therefore, better conversion and selectivity towards gasoline range hydrocarbons.

Materials and Methods

Fe₃O₄ nanocatalysts have been synthetized by a new ionic liquid-assisted synthesis by heating the reaction medium consisting of the ionic liquid 1-butyl-3-methyl imidazolium bistriflimide, $[C_4mim][Tf_2N]$, oleic acid and iron pentacarbonyl under reflux (Method 1). The precursor iron pentacarbonyl decomposed in a controlled manner by heating the sample up; CO is produced and the iron reacts with residual H₂O in the ionic liquid mixture to result in Fe₃O₄. Following decomposition, the produced magnetite nanoparticles are separated from the reaction medium through application of a neodymium magnet.

Another ionic liquid-assisted synthesis method consisted of heating two iron precursors of Fe (II) and Fe (III) chloride hydrates in a reaction medium of $[C_4mim][OAc]$ ionic liquid (Method 2) w/o ammonia. This method involves calcination of the iron oxide nanoparticles at 420°C under N₂ to prepare crystalline material. The Fe₃O₄ particles were then supported on zeolites by granule mixing Fe₃O₄ particles prepared with the methods above with zeolite HZSM-5 (SiO₂/Al₂O₃ = 300) in a ball mill at a mass ratio of the two components of 1:1.

 CO_2 hydrogenation reactions were performed at 320 °C, 3 MPa H /CO 3,3 in a stainless steel fixed-bed reactor with an inner diameter of 15 mm. Typically,1 g of composite catalyst (20–40 meshes) with Fe₃O₄/Zeolite 1/4 1/1 (mass ratio) was used. Prior to reaction, the catalyst was insitu reduced at 350 °C for 8 h in a pure H₂ flow at atmospheric pressure.

All of the products from the reactor were introduced in a gaseous state and analysed with an online gas chromatograph (GC).

Results and Discussion

The ionic-liquid assisted synthesis of a nanocrystalline magnetite precursor showed that ionic liquids provide a controlled precipitation method thanks to their dual functionality as solvent and templating agent. This level of control over the morphology of the produced IONs allows for the selectivity of the hydrocarbon distribution to be directly tailored. Characterization of the

prepared catalysts by PXRD (Figure 1) shows the presence of high purity Fe₃O₄, small particle size and good dispersion with the zeolite component for both Method 1 and 2. This has also been confirmed by SEM and TEM analysis. Hydrogen temperature-programmed reduction (H₂-TPR) was used to determine the reducibility of the Fe₃O₄ particles and their hydrogen uptake. As shown in Figure 2, all the catalysts presented two peaks of H₂ consumption, which are assigned to the conversions Fe₃O₄—FeO and FeO—Fe, respectively. It is observed that Fe₃O₄/HZSM-5 (Method 1) is reduced at lower temperature compared to the catalysts prepared with Method 2. All catalysts prepared with ionic liquid-assisted synthesis showed high reducibility in the low temperature region (250-350°C), which correspond to the activation temperature typical of Fischer -Tropsch catalysts.



Figure 1. XRD patterns of ionic liquid methods.

Figure 2. TPR profile of 5%H₂ in Ar from RT to700°C, 10°C/min

Significance

We report here on a novel methodology for the controlled synthesis of a Fe₃O₄/HZSM-5 multifunctional catalyst for the direct hydrogenation of CO₂ to gasoline. The catalyst morphology can be tuned by the choice of ionic liquid in the synthetic method and this also affect the selectivity of the reaction. The catalytic testing under industrially relevant conditions resulted in improved selectivity to C5–C11 as well as low CH₄ and CO selectivity

This study provides a new pathway for the synthesis of nanocatalysts to produce liquid fuels by utilising CO_2 and H_2 , which may in the future lead to alternative approaches to overcome issues with the intermittency of storing and/or utilising energy from renewable sources (photovoltaics, wind energy).

References

¹Y. Yuan, S. Huang, H. Wang, Y. Wang, J. Wang, J. Lv, Z. Li, and X. Ma, ChemCatChem 2017, 9, 3144 – 3152

² J. Wei, Q. Ge, R. Yao, Z. Wen, C. Fang, L. Guo, H. Xu, J. Sun, Nat Comm, DOI: 10.1038/ncomms15174



| Tuesday, 7 th January | | | | |
|----------------------------------|---|--------------------------------------|------------------------------|--|
| 11:00 | 11:00 Registration desk opens at Burleigh Court Hotel | | | |
| 12:30 | Lunch at Holywell Park | | | |
| 13.50 | Welcome – Conference commence | s at Holywell Park | | |
| | | Chair - Catlow | | |
| 14.00 | | Duncan Wass (Turing Lecture Theatre) | | |
| 14.45 | | Coffee | | |
| | Session A | Session B | Session C | |
| | (Turing Lecture Theatre) | (Brunel/Murdoch Lecture Theatre) | (Stephenson Lecture Theatre) | |
| | CatalysisHub session | | | |
| Chair/IT | Garforth/Deshmukh | Lennon/Shiels | Diez-Gonzalez/Keogh | |
| 15.15 | K1 | O22 | O47 | |
| 15.35 | | O23 | O48 | |
| 15.55 | O1 | O24 | O49 | |
| 16.15 | O2 | O25 | K9 | |
| 16.35 | O3 | O26 | | |
| 16.55 | Coffee | | | |
| Chair/IT | Taylor/Keogh | Kondrat/McDermott | Marr/Isah | |
| 17.25 | O4 | K6 | O50 | |
| 17.45 | O5 | | O51 | |
| | Chair - Hardacre | | | |
| 18.10 | Johannes Lercher (Turing Lecture Theatre) | | | |
| 20.00 | Dinner | | | |



| Wednesday, 8 th January | | | | |
|------------------------------------|---|---|---|--|
| | Chair - Hutchings | | | |
| 9.00 | | Angelika Brückner (Turing Lecture Theatre |) | |
| | Session A (Turing Lecture Theatre) | Session B (Brunel/Murdoch Lecture Theatre) | Session C (Stephenson Lecture Theatre) | |
| Chair/IT | McGregor/Sun | Fan/McDermott | Wood/Tanvir | |
| | RSC INTEREST GROUP SURFACE REACTIVITY SESSION & CATALYSIS | | | |
| 9.50 | K2 | O27 | O52 | |
| 10.10 | | O28 | O53 | |
| 10.30 | O6 | O29 | O54 | |
| 10.50 | | Coffee | | |
| Chair/IT | Thompson/Akor | Wu/Keogh | Reina/Hao | |
| 11.20 | K3 | O30 | O55 | |
| 11.40 | | O31 | O56 | |
| 12.00 | 07 | O32 | O57 | |
| 12.20 | O8 | K7 | O58 | |
| 12.40 | O9 | | O59 | |
| 13.00 | | Lunch | | |
| | | Chair - Manyar | | |
| 14.00 | ALMAL | José Odriozola (Turing Lecture Theatre) | ALMAL: | |
| 14.45 | | Coffee | | |
| | Session A | Session B | Session C | |
| | (Turing Lecture Theatre) | (Brunel/Murdoch Lecture Theatre) | (Stephenson Lecture Theatre) | |
| Chair/IT | Paterson/Yue | Moody/McDermott | Whiston/Deshmukh | |
| | C RSC INTEREST GROUP SESSION | | | |
| 15.15 | K4 | K8 | O60 | |
| 15.35 | | | O61 | |
| 15.55 | O10 | O33 | O62 | |
| 16.15 | O11 | O34 | K10 | |
| 16.35 | O12 | O35 | | |
| 16.55 | Coffee | | | |

| 2020 | | | Loughborough, UK |
|-----------------------------------|---|----------------------------------|------------------------------|
| Chair/IT | Kroner/Shiels | Berlier/Sun | Raveendran/Akor |
| 17.25 | O13 | O36 | O63 |
| 17.45 | O14 | O37 | O64 |
| 18.05 | O15 | O38 | O65 |
| 18.30 | BP Poster session | | |
| 20.00 | | Conference Dinner | |
| Thursday, 9 th January | | | |
| | Session A | Session B | Session C |
| | (Turing Lecture Theatre) | (Brunel/Murdoch Lecture Theatre) | (Stephenson Lecture Theatre) |
| Chair/IT | Beale/Keogh | Artioli/Deshmukh | Upadhyayula/Tanvir |
| 9.00 | O16 | O39 | O66 |
| 9.20 | 017 | O40 | K11 |
| 9.40 | O18 | O41 | |
| 10.00 | K5 | O42 | O67 |
| 10.20 | | O43 | O68 |
| 10.40 | Coffee | | |
| Chair/IT | Minova/Shiels | Mitchell/Deshmukh | Hintermair/McDermott |
| 11.10 | O19 | O44 | K12 |
| 11.30 | O20 | O45 | |
| 11.50 | O21 | O46 | O69 |
| | Chair - Davidson | | |
| 12.20 | Stewart Parker (Turing Lecture Theatre) | | |
| 13.05 | Closing remarks | | |







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List of Talks UKCC 2020

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| | Feedstocks to Advanced Synthetic Fuels | | |
| PI 02 | New strategies for enhancing catalytic | Johannes Lercher | |
| | rates | | |
| PI 03 | Identifying active sites and mechanisms: | Angelika Brückner | |
| | Opportunities and limitations of in situ and | | |
| | operando spectroscopy in catalysis | | |
| PI 04 | From electrons to reactors: The WGS revisited | Jose Odriozola | |
| PI 05 | What's on your catalyst? Characterization | Stewart Parker | |
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| K 01 | Elementary Steps in the Formation of | Ivalina Minova, Santhosh Matam, Alex | |
| | Olefins from Surface Methoxy Groups in | Greenaway, Richard Catlow, Mark | |
| | ZSM-5 and SAPO-34 Seen by Operando | Frogley, Gianfelice Cinque, Paul Wright | |
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| K US | Reduction: insights into Cu speciation and | | |
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| | spectroscopic techniques | | |
| K 04 | Insights into the CO ₂ formation pathways | Shashank Bahri and Sreedevi | |
| | over bimetallic Fischer-Tropsch catalyst for | Upadhyayula | |
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| K 05 | Catalytic scissoring of lignin C-C and C-O | Wang, Luo, Liu and Li | |
| | bonds | | |
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| K 07 | Catalyst | | |
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| K 08 | (Trans)Forming C-N and C-O Bonds with | Silvia Díez-González | |
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| К 11 | MAX Phases and MXenes as Efficient | Shiju Bayeendran |
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| | | and Adam Lee |
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| 0 27 | CeFeO _x catalysts for the total oxidation of | Kieran Aggett and Stuart Taylor |
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| O 32 | A heterogeneous platform for biocatalytic asymmetric deuteration | Jack Rowbotham, Miguel Ramirez Hernandez, Oliver Lenz, Holly Reeve and Kylie Vincent |
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